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# .Synthesis of New 1,3-Oxazole Derivatives 

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#### Abstract

1,3- Oxazole derivatives ( $2 \mathrm{a}-\mathrm{d}$ ) have been Synthesized by cyclization of hippuric acid with different aromatic aldehydes, hippuric acid was readily obtained by reaction of benzoyl chloride with glycine ( 2 -amino acetic acid). Compounds ( $2 a-d$ ) were converted in to aviary of derivatives. All new compounds were characterized by H NMR, FTIR and UV spectroscopy

الخلاصة تم في هذا البحث تحضير مركبات (2a-d) من خلال اجراء عملية الغلق الحلقي لحامض الهبيوريك مع الديهايدات اروماتية مختلفة بينما حضر المركب الاساسي حامض الهبيوريك من تفاعل المركب benzoyl chloride مع الحامض الاميني (Glycine) ثٌ بعد ذلك تم تحويل المركبات (2a-d) الى بقية المشتقات ـ كل المركبات المحضرة تم

H NMR, FTIR and UV تثخيصها بأستخدام الطر ائق الطيفية


## INTRODUCTION

Oxazoles are a kind of attractive heterocycles not only because of their unique structures and varied applications but also they serve as structural elements for a variety of natural products[1], pharmaceuticals[2] and bioactive compounds[3], A number of synthetic methods to prepare oxazoles have been reported[4]. The typical procedure for the synthesis of oxazoles involves the reaction of readily available substituted urea derivatives with halogenated alkenes or $\alpha$-haloketones [5]. Imidazoles are an important class of heterocycles being the core fragment of different natural products and biological systems[6], Compounds containing imidazole moiety have many pharmacological properties and play important roles in biochemical processes [7]. Imidazolones have been associated with several pharmacological activities such as antimicrobial (antifungal, antibacterial and antiviral), anticancer activity, CNS depressant activity[8]. Chalcones are an important class of compounds which are good intermediates for the synthesis of various heterocyclic compounds like flavones, isoxazolines, pyrimidines, quinoxalines, benzalcoumaranones [9] .Chalcones display a wide range of pharmacological properties, including cytotoxity towards cancer cell lines[10]. Pyrimidine is the most important six membered heterocyclic containing 2 nitrogen atoms at position 1 and 3. It is isomeric with two other forms of diazene[11]. Pyrazole is a five-membered heterocyclic moiety having two adjacent nitrogen atoms within the ring, and It is basic in nature [12], pyrazole is one of the most important one as large variety of biological activities have been reported for various pyrazole derivatives. Conventional method of synthesis of pyrazoles involves the base-catalyzed condensation of aromatic ketones to give $\alpha, \beta$ - unsaturated ketones (also called as chalcones) , which undergo subsequent cyclization with hydrazine and hydrazine derivatives[13,14]. The pyrazole nucleus is a ubiquitous
feature of pharmacological interest and has been proven to be a fertile source of medicinal agents , pyrazole derivatives have also exhibited antidiabetic properties ,and some of these have biological activities such as anti-inflammatory[15,16].

## MATERIALS AND METHODS


(10a,10b)
$\mathrm{Ar}=4-\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~N}-\mathrm{C}_{6} \mathrm{H}_{4}, 4-\mathrm{Br}-\mathrm{C}_{6} \mathrm{H}_{4}, 4-\mathrm{Cl}-\mathrm{C}_{6} \mathrm{H}_{4}, 2-\mathrm{NO}_{2}-\mathrm{C}_{6} \mathrm{H}_{4}$
Scheme(1)
Synthesis of 2-benzamidoacetic acid (1)[17]
To a mixture of glycine ( 0.18 mole , 13.2 gm ) in mixture of sodium hydroxide ( 60 ml ) $10 \% \mathrm{NaOH}$ was added and stirred for 10 min ,then benzoyl chloride ( 0.18 mole , 25 gm ) was added to the mixture,then the reaction mixture was allowed to cool then acidified with cons HCl , ice cooled water was added to the solid product obtained and filtered and recrystallized from apposite solvent.
(1) :Yield, $74 \%$, m.p. 184-185 $\mathrm{C}^{\circ}$, $\mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3342$ $\mathrm{v}(\mathrm{NH}), 2484-3400 \mathrm{v}(\mathrm{O}-\mathrm{H}) 3070 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2893,2939 v(C-H)aliphatic , $1753 \quad v(\mathrm{C}=\mathrm{O})$ acid ,
$1687 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide. $\mathrm{U} . \mathrm{V}(\mathrm{MeOH}): 229 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right)$, $209 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.


$\mathrm{Ar}=4-\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~N}-\mathrm{C}_{6} \mathrm{H}_{4}, 4-\mathrm{Br}-\mathrm{C}_{6} \mathrm{H}_{4}, 4-\mathrm{Cl}-\mathrm{C}_{6} \mathrm{H}_{4}, 2-\mathrm{NO}_{2}-\mathrm{C}_{6} \mathrm{H}_{4}$ Scheme(ll)

## Synthesis of (E)-4-(Arylidene)-2-phenyl-1,3- oxazol-5(4H)-ones(2a-d)[18]

A mixture of (1) ( 0.02 mole, 3.58 gm ) and aromatic aldehydes ( 0.02 mole) in acetic anhydride ( 20 ml ) and acetic acid ( 5 ml )was refluxed for 3 hrs , then the reaction mixture was added to ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$2_{\mathrm{a}}$ : Yield, $89 \%$, m.p. 214-215 C ${ }^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3016$ v(CH)aromatic, 2812,2899 $\mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1784 \mathrm{v}(\mathrm{C}=\mathrm{O})$ lactone , $1645 v(\mathrm{C}=\mathrm{N})$, $(1606,1581) \quad v(\mathrm{C}=\mathrm{C})$ ar . U.V $(\mathrm{MeOH}): 301 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 242 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$2_{\mathrm{b}}$ : Yield, $75 \%$,m.p. $154-156 \mathrm{C}^{\circ}, \mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3057 \mathrm{v}(\mathrm{C}-$ H)aromatic, 2839,2985 $\mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1795 \mathrm{v}(\mathrm{C}=\mathrm{O})$ lactone, $1653 v(\mathrm{C}=\mathrm{N})$, $(1583,1554) \mathrm{v}(\mathrm{C}=\mathrm{C})$ ar . U.V (MeOH) : $388 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 252 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$2_{\mathrm{c}}$ : Yield, $83 \%$, m.p. 116-118 C ${ }^{\circ}$, IR( $\left.\mathrm{cm}^{-1}\right): 3078$ v(CH)aromatic, 2847,2991 $\mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1795 \mathrm{v}(\mathrm{C}=\mathrm{O})$ lactone, $1653 v(\mathrm{C}=\mathrm{N}),(1556,1589) \mathrm{v}(\mathrm{C}=\mathrm{C})$ ar . U.V (MeOH) : $348 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 232 \mathrm{~nm}\left(\pi-\pi^{*}\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) (ppm) 7.38 ( $\mathrm{s},=\mathrm{CH}$ - alkene) , 7.60-7.77 ( $\mathrm{s}, 5 \mathrm{H}$ for phenyl ring ), 8.13-8.34 (dd, 4 H for aryl ring).
$2_{\mathrm{d}}$ : Yield, $70 \%$, m.p. $165-167 \mathrm{C}^{\circ}$, IR $\left(\mathrm{cm}^{-1}\right): 3070$ v(CH)aromatic, 2839,2985 $v(\mathrm{C}-\mathrm{H})$ aliphatic ,1797 $\mathrm{v}(\mathrm{C}=\mathrm{O})$ lactone, $1654 v(\mathrm{C}=\mathrm{N})$, $(1550,1599) \mathrm{v}(\mathrm{C}=\mathrm{C})$ ar . U.V $(\mathrm{MeOH}): 362 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 245 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.

Synthesis of (E)-1-(4-(Arylidene)-5-oxo-2-phenyl-4,5-dihydro-1H-imidazol-1-yl)thiourea(3a-d)[19]
A mixture of ( $2 \mathrm{a}-\mathrm{d}$ ) ( 0.02 mole ) and thiosemicarbazide ( 0.02 mole , $\quad 0.18 \mathrm{gm}$ ) in pyridine ( 20 ml ) was refluxed for 20 hrs , then the reaction mixture was added to ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$3_{\mathrm{a}}$ : Yield, $76 \%$, m.p. $192-194 \mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3412,3466$ $\mathrm{v}\left(\mathrm{NH}_{2}\right)$, $3075 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2858,2906 $\mathrm{v}(\mathrm{C}-$ H) aliphatic , $1693 \mathrm{v}(\mathrm{C}=\mathrm{O})$ lactam , $1633 \mathrm{v}(\mathrm{C}=\mathrm{N})$.
$3_{\mathrm{b}}$ : Yield, $70 \%$, m.p. $145-147 \mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right)$ : 3317,3336 $\mathrm{v}\left(\mathrm{NH}_{2}\right)$, $3070 \quad \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2870,2928 $\mathrm{v}(\mathrm{C}-$ H)aliphatic, $1708 \mathrm{v}(\mathrm{C}=\mathrm{O})$ lactam , $1631 \mathrm{v}(\mathrm{C}=\mathrm{N})$. U.V (MeOH) : $389 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 272 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$3_{\mathrm{c}}$ : Yield, $67 \%$, m.p. $135-137 \mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3400,3439$ $\mathrm{v}\left(\mathrm{NH}_{2}\right)$, $3073 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2852,2926 $\mathrm{v}(\mathrm{C}-$ H)aliphatic, $1710 v(\mathrm{C}=\mathrm{O})$ lactam, $1645 \mathrm{v}(\mathrm{C}=\mathrm{N})$.
$3_{\mathrm{d}}$ : Yield, $74 \%$, m.p. $165-167 \mathrm{C}^{\circ}$, IR $\left(\mathrm{cm}^{-1}\right)$ : 3375,3444 $\mathrm{v}\left(\mathrm{NH}_{2}\right) 3068$ (C-H)aromatic, 2870,2920 v(CH)aliphatic, $1707 \mathrm{v}(\mathrm{C}=\mathrm{O})$ lactame, $1641 \mathrm{v}(\mathrm{C}=\mathrm{N})$. U.V $(\mathrm{MeOH}): 315 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 266 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.

## Synthesis of 3-(Aryl)-5-phenyl-1,3-dihydrooxazolo [5,4-c] isoxazol (4a-d)[20]

A mixture of (2a-d) ( 0.001 mole) and hydroxyl amine hydrochloride ( 0.003 mole, 0.2 gm ) and sodium acetate anhydrous ( 0.003 mole , 0.24 gm ) in ( 15 ml ) ethanol and $(5 \mathrm{ml})$ glacial acetic acid was refluxed for 8 hrs , then the reaction mixture was added to ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$4_{\mathrm{a}}$ : Yield, $70 \%$, m.p. 147-148 $\mathrm{C}^{\circ}, \operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3201 \mathrm{v}(\mathrm{NH})$ , 2858,2912 v(C-H)aliphatic, $1670 v(\mathrm{C}=\mathrm{N}), 1516,1587$ $\mathrm{v}(\mathrm{C}=\mathrm{C})$ aromatic . U.V (MeOH) : $472 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 281 \mathrm{~nm}$ ( $\pi-\pi^{*}$ ).
$4_{b}$ : Yield, $84 \%$, m.p. $160-162 \mathrm{C}^{\circ}$, IR( $\mathrm{cm}^{-1}$ ): $3151 \mathrm{v}(\mathrm{NH})$ , 2816,2912 v(C-H)aliphatic, $1703 v(\mathrm{C}=\mathrm{N}), 1560,1583$ $\mathrm{v}(\mathrm{C}=\mathrm{C})$ aromatic . U.V (MeOH) : $387 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 262 \mathrm{~nm}$ ( $\pi-\pi^{*}$ ) . ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) (ppm) 7.02(-CH-) isoxazole ring, $7.58-8.28$ ( 5 H s for phenyl ring ,4 H dd for aryl ring ), $11.34(1 \mathrm{H} \mathrm{s}, \mathrm{NH})$.
$4_{c}$ : Yield, $62 \%$, m.p. $170-172 \mathrm{C}^{\circ}, \operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3180 \mathrm{v}(\mathrm{NH})$ , 2850,2918 v(C-H)aliphatic, $1701 \mathrm{v}(\mathrm{C}=\mathrm{N}) . \mathrm{U} \cdot \mathrm{V}(\mathrm{MeOH})$ $: 380 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 238 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$4_{\mathrm{d}}$ : Yield, $57 \%$, m.p. $128-130 \mathrm{C}^{\circ}, \operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3178 \mathrm{v}(\mathrm{NH})$ , 2870,2987 $v(\mathrm{C}-\mathrm{H})$ aliphatic, $1716 \mathrm{v}(\mathrm{C}=\mathrm{N})$.

## Synthesis of 7-(Aryl)-2-phenyl-6,7-dihydrooxazolo

 [5,4-d] pyrimidine-5(4H)-one (5a-d)[21]A mixture of ( $2 \mathrm{a}-\mathrm{d}$ ) ( 0.002 mole ) and urea ( 0.002 mole , 0.12 gm ) was dissolved in a mixture of acetone ( 15 ml ) and ethanol ( 15 ml ),then potassium carbonate ( 0.002 mole , 0.27 gm ) was added with vigorous stirring then the mixture was refluxed for 14 hrs , after this the reaction mixture was poured in ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$5_{\mathrm{a}}$ : Yield, $86 \%$, m.p. $144-145 \mathrm{C}^{\circ}, \mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3309 \mathrm{v}(\mathrm{NH})$ , 2812,2978 v(C-H)aliphatic , $1707 \mathrm{v}(\mathrm{C}=\mathrm{O}), 1645$
$v(\mathrm{C}=\mathrm{N})$ cyclic . U.V (MeOH) : $365 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 234 \mathrm{~nm}($ $\pi-\pi^{*}$ ).
$5_{\mathrm{b}}$ : Yield, $58 \%$, m.p. $179-181 \mathrm{C}^{\circ}, \operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3211 \mathrm{v}(\mathrm{NH})$ , 2890,2978 v(C-H) aliphatic , $1716 \mathrm{v}(\mathrm{C}=\mathrm{O}), 1649$
$\mathrm{v}(\mathrm{C}=\mathrm{N})$ cyclic. U.V(MeOH): $358 \mathrm{~nm} \quad\left(\mathrm{n}-\pi^{*}\right), 292 \mathrm{~nm}($ $\pi-\pi *$ )
$5_{\mathrm{c}}$ : Yield, $75 \%$, m.p. $50-52 \mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3209 \mathrm{v}(\mathrm{NH})$, 2899,2976 $v(\mathrm{C}-\mathrm{H})$ aliphatic , $1718 v(\mathrm{C}=\mathrm{O}), 1649$ $\mathrm{v}(\mathrm{C}=\mathrm{N})$ cyclic. $\mathrm{U} . \mathrm{V}(\mathrm{MeOH}): 289 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 228 \mathrm{~nm}(\pi-$ $\pi^{*}$ ).
$5_{\mathrm{d}}$ : Yield, 57\% , m.p. 150-152 C ${ }^{\circ}$, IR( $\left.\mathrm{cm}^{-1}\right): 3221 \mathrm{v}(\mathrm{NH})$ , 2812,2985 $\mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1726 \mathrm{v}(\mathrm{C}=\mathrm{O}), 1651$ $v(\mathrm{C}=\mathrm{N})$ cyclic.

## Synthesis of 3-(Aryl)-5-phenyl-2,3-dihydro-1Hpyrazolo [4,3-d] oxazole(6a-d)[22]

A mixture of (2a-d) ( 0.005 mole) and hydrazine hydrate ( 0.005 mole , 0.25 gm ) in ( 20 ml ) acetic acid was refluxed for 8 hrs , then the reaction mixture was added to ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$6_{\mathrm{a}}$ : Yield, $78 \%$, m.p. 170-172 $\mathrm{C}^{\circ}$, $\mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3130$ $\mathrm{v}(\mathrm{NH})$ far from oxazole ring , $3200 \mathrm{v}(\mathrm{NH})$ near from oxazole ring , 2848,2943 v(C-H)aliphatic , 1662 $v(\mathrm{C}=\mathrm{N})$ cyclic .U.V(MeOH) : $305 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 259 \mathrm{~nm}(\pi-$ $\pi^{*}$ ).
$6_{b}$ : Yield, $68 \%$, m.p. 211-213 $\mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3201$ $\mathrm{v}(\mathrm{NH})$ far from oxazole ring , $3311 \mathrm{v}(\mathrm{NH})$ near from oxazole ring, 2893,2953 $\mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1658 \mathrm{v}(\mathrm{C}=\mathrm{N})$ cyclic. U.V(MeOH) :259 nm(n- $\left.\pi^{*}\right)$, $228 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$6_{\mathrm{c}}$ : Yield, $71 \%$, m.p. 126-128 $\mathrm{C}^{\circ}$, $\mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3130$ $\mathrm{v}(\mathrm{NH})$ far from oxazole ring , $3209 \mathrm{v}(\mathrm{NH})$ near from oxazole ring, 2852,2960 $\mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic, $1654 \mathrm{v}(\mathrm{C}=\mathrm{N})$ cyclic. U.V(MeOH) :284 nm (n- $\left.\pi^{*}\right), 228 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$6_{\mathrm{d}}$ : Yield, $76 \%$, m.p. 208-210 $\mathrm{C}^{\circ}$, $\mathrm{IR}\left(\mathrm{cm}^{-1}\right): 2211$ $\mathrm{v}(\mathrm{NH})$ far from oxazole ring , $3258 \mathrm{v}(\mathrm{NH})$ near from oxazole ring , $2812,2985 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1726 \mathrm{v}(\mathrm{C}=\mathrm{O})$, $1664 v(\mathrm{C}=\mathrm{N})$ cyclic. $\mathrm{U} . \mathrm{V}(\mathrm{MeOH}): 258 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 224 \mathrm{~nm}$ ( $\pi-\pi *$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) (ppm) 4.63 (s ,NH far from oxazole ring and -CH-) , 7.40-8.26 (5H s for phenyl ring, 4 H dd for aryl ring ), 13.59 ( $\mathrm{s}, \mathrm{NH}$ near from oxazole ring).

Synthesis of Ethyl 2-(3-(Aryl)-5-phenyl-1H-pyrazolo [4,3-d] oxazol -2 (3H)-yl)acetate (7a-d) [23]
A mixture of ( $6 \mathrm{a}-\mathrm{d}$ ) ( 0.003 mole) and ethyl chloro acetate ( 0.003 mole, 0.36 gm ) and sodium carbonate ( $0.003 \mathrm{~mole}, 0.31 \mathrm{gm}$ ) in ethanol ( 20 ml ) was refluxed for 7 hrs , then the reaction mixture was added to ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$7_{\mathrm{a}}$ : Yield, $62 \%$, m.p. $153-155 \mathrm{C}^{\circ}, \mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3130 \mathrm{v}(\mathrm{NH})$ , $3003 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2854,2970 v(C-H)aliphatic , 1734 $v(\mathrm{C}=\mathrm{O}), 1658 v(\mathrm{C}=\mathrm{N}) . \mathrm{U} \cdot \mathrm{V}(\mathrm{MeOH}): 306 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 260$ $\mathrm{nm}\left(\pi-\pi^{*}\right)$.
$7_{\mathrm{b}}$ : Yield, $71 \%$, m.p. $133-135 \mathrm{C}^{\circ}$, $\mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3078 \mathrm{v}(\mathrm{NH})$ , $3051 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2843,2939 v(C-H)aliphatic , 1747 $v(\mathrm{C}=\mathrm{O}), 1658 v(\mathrm{C}=\mathrm{N}) . \mathrm{U} \cdot \mathrm{V}(\mathrm{MeOH}): 269 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 253$ $\mathrm{nm}\left(\pi-\pi^{*}\right)$.
$7_{\mathrm{c}}$ : Yield, $60 \%$, m.p. $123-125 \mathrm{C}^{\circ}, \operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3203 \mathrm{v}(\mathrm{NH})$ , $3073 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2864,2978 v(C-H)aliphatic , 1737 $v(\mathrm{C}=\mathrm{O}), 1664 v(\mathrm{C}=\mathrm{N}) . \mathrm{U} \cdot \mathrm{V}(\mathrm{MeOH}): 229 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 214$ $\mathrm{nm}\left(\pi-\pi^{*}\right)$.
$7_{\mathrm{d}}$ : Yield, $71 \%$, m.p. $100-102 \mathrm{C}^{\circ}$, IR( $\left.\mathrm{cm}^{-1}\right): 3080 \mathrm{v}(\mathrm{NH})$ , $3001 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2912,2951 v(C-H)aliphatic , 1749 $v(\mathrm{C}=\mathrm{O}), 1656 v(\mathrm{C}=\mathrm{N})$. U.V(MeOH): $306 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 260$ $\mathrm{nm}\left(\pi-\pi^{*}\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) (ppm) 1.21-1.25 (triplet $\mathrm{CH}_{3}$ ) , $4.17-4.24$ (quartate, $-\mathrm{CH}_{2}$ - $\left.\mathrm{O}-\right), 4.68\left(-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right)$ , 4.98 ( $\mathrm{s}, \mathrm{NH}$ and $-\mathrm{CH}-$ ), 7.42-8.28 ( 5 H s for phenyl ring , 4 H dd for aryl ring) .

Synthesis of 2-(3-(Aryl)-5-phenyl-1H-pyrazolo[4,3-d]oxazol-2(3H)-yl)acetohydrazide(8a-d)[24]
A mixture of (7a-d) ( 0.002 mole) and hydrazine hydrate ( 0.002 mole , 0.1 gm ) in ethanol ( 20 ml ) was refluxed for 10 hrs , then the reaction mixture was added to ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$8_{\mathrm{a}}$ : Yield, $56 \%$, m.p. 187-189 C ${ }^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3252,3288$ $\mathrm{v}\left(\mathrm{NH}_{2}\right)$, $3034 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2850,2920 v(C-H) aliphatic, $1651 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide . U.V (MeOH) : $304 \mathrm{~nm}($ $\left.\mathrm{n}-\pi^{*}\right), 244 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$8_{\mathrm{b}}$ : Yield, $77 \%$,m.p. 182-184 $\mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3290,3296$ $\mathrm{v}\left(\mathrm{NH}_{2}\right)$, $3070 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, $2854,2929 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic, $1658 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide . U.V (MeOH) : $228 \mathrm{~nm}($ $\left.\mathrm{n}-\pi^{*}\right), 213 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$8_{\mathrm{c}}$ : Yield, $60 \%$, m.p. $128-130 \mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3294,3317$ $\mathrm{v}\left(\mathrm{NH}_{2}\right), 3067 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2848,2916 v(C-H) aliphatic, $1629 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide $. \mathrm{U} . \mathrm{V}(\mathrm{MeOH}): 385 \mathrm{~nm}(\mathrm{n}-$ $\left.\pi^{*}\right), 236 \mathrm{~nm}\left(\pi-\pi^{*}\right)$. Mass $\operatorname{Spectra}(367 \mathrm{~g} /$ mole $)$.
$8_{\text {d }}$ : Yield, 64\% , m.p. 178-180 $\mathrm{C}^{\circ}$,IR( $\left(\mathrm{cm}^{-1}\right): 3311,3319$ $\mathrm{v}\left(\mathrm{NH}_{2}\right)$, $3061 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2848,2918 $v(\mathrm{C}-\mathrm{H})$ aliphatic, $1683 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide . U.V (MeOH) : 260 nm ( $\left.\mathrm{n}-\pi^{*}\right), 226 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.

## Synthesis of (E)-N-(4-bromobenzyl)-2-(3-(Aryl)-5-phenyl-1H-pyrazolo <br> [4,3-d]oxazol-2(3H)yl)acetohydrazide (9a-d)[25]

A mixture of aromatic aldehydes ( 0.001 mole) was dissolved in ( 20 ml ) ethanol with ( 3 drops) from glacial acetic acid, then added ( 0.001 mole, 0.41 gm ) from the compounde ( 8 b ) to the mixture and refluxed for 6 hrs , then the reaction mixture was added to ice water and the solid product obtained was filtered and recrystallized from apposite solvent.
$9_{\mathrm{a}}$ : Yield, $75 \%$, m.p. $135-136 \mathrm{C}^{\circ}$, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3244 \mathrm{v}(\mathrm{NH})$
$3059 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, $2854,2922 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1681 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide , $1666 \mathrm{v}(\mathrm{C}=\mathrm{N})$. U.V $(\mathrm{MeOH}): 341$ nm ( $\mathrm{n}-\pi^{*}$ ), $229 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$9_{\mathrm{b}}$ : Yield, $70 \%$,m.p. $295 \mathrm{C}^{\circ}$ decompostion, $\operatorname{IR}\left(\mathrm{cm}^{-1}\right)$ : $3134 v(\mathrm{NH})$, $3066 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2852,2924v(C-H) aliphatic , $1660 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide , $1629 \mathrm{v}(\mathrm{C}=\mathrm{N})$. U.V (MeOH) : $288 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 227 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$9_{\mathrm{c}}$ : Yield, $68 \%$, m.p. $173-175 \mathrm{C}^{\circ}, \operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3130 v(\mathrm{NH})$, $3049 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, $2881,2929 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic, 1681 $v(\mathrm{C}=\mathrm{O})$ amide , $1624 \mathrm{v}(\mathrm{C}=\mathrm{N})$. U.V (MeOH) : $260 \mathrm{~nm}($ $\left.\mathrm{n}-\pi^{*}\right), 227 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$9_{\mathrm{d}}$ : Yield, $68 \%$, m.p. 268-270 $\mathrm{C}^{\circ}$, IR $\left(\mathrm{cm}^{-1}\right): 3213 \mathrm{v}(\mathrm{NH})$ , $3068 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2852,2924 $\mathrm{v}(\mathrm{C}-\mathrm{H})$ aliphatic , $1654 \mathrm{v}(\mathrm{C}=\mathrm{O})$ amide , $1639 \mathrm{v}(\mathrm{C}=\mathrm{N})$. U.V (MeOH) : 260 $\mathrm{nm}\left(\mathrm{n}-\pi^{*}\right), 227 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.

Synthesis of (E)-4-(Arylidene)-1-(5-oxo-2-thioxoimidazolidin-1-yl)-2-phenyl-1H-imidazol-5(4H)one ( $10 \mathrm{a}, \mathrm{b}$ )[23]
A mixture of (3a,b) ( 0.001 mole) and ethyl chloro acetate ( 0.001 mole , 0.122 gm ) in ethanol ( 20 ml ) was refluxed for 10 hrs ,then the reaction mixture was added to ice water and the solid product obtainedwas filtered and recrystallized from apposite solvent.
$10_{\mathrm{a}}$ : Yield, $76 \%$, m.p. $123-125 \mathrm{C}^{\circ}, ~ \mathrm{IR}\left(\mathrm{cm}^{-1}\right): 3300$ $\mathrm{v}(\mathrm{NH})$, $3086 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2858,2910 v(C-H) aliphatic , $1720 \mathrm{v}(\mathrm{C}=\mathrm{O})$ New imidazole ring . U.V (MeOH) : $354 \mathrm{~nm}\left(\mathrm{n}-\pi^{*}\right), 269 \mathrm{~nm}\left(\pi-\pi^{*}\right)$.
$10_{\mathrm{b}}$ : Yield, $72 \%$,m.p. $130-132 \mathrm{C}^{\circ}, \operatorname{IR}\left(\mathrm{cm}^{-1}\right): 3318 \mathrm{v}(\mathrm{NH})$ , $3077 \mathrm{v}(\mathrm{C}-\mathrm{H})$ aromatic, 2850,2949 v(C-H) aliphatic, $1714 \mathrm{v}(\mathrm{C}=\mathrm{O})$ New imidazole ring . Mass Spectra (442 g /mole ) .

## RESULTS AND DISCUSSION

Schemes (I - II) were summarized the synthesis of different derivatives of hippuric acid (1) which was synthesized by treatment of benzoyl chloride with glycine. The reaction is followed by the appearance of the new ( $\mathrm{C}=\mathrm{O}$ ) band at $\left(1687 \mathrm{Cm}^{-1}\right)$ due to carbonyl of amide and band at 1753 for ( $\mathrm{C}=\mathrm{O}$ acid) and bands at (2484-3400 $\mathrm{Cm}^{-1}$ ) for stretching vibration of (acidic OH ). $\lambda_{\text {max }}$ $(\mathrm{MeOH})$ at $(229 \mathrm{~nm})$ responsible for $\left(\mathrm{n} \rightarrow \pi^{*}\right)$ transition of ( N and O ) atoms and at ( 209 nm ) due to ( $\pi \rightarrow \pi^{*}$ ). Compounds ( $2 \mathrm{a}-\mathrm{d}$ ) have been synthesized by the reaction of compound (1) with different aromatic aldehydes in acetic anhydride and acetic acid, the reaction proceeds by elimination of $\mathrm{H}_{2} \mathrm{O}$ molecule. The reaction is followed by appearance of the new $(\mathrm{C}=\mathrm{O})$ band at $\left(1784-1797 \mathrm{Cm}^{-1}\right)$ for lactone ring which showed the increase of frequency of carbonyl and band at (1645-1654 $\mathrm{Cm}^{-1}$ ) for stretching vibration of $(\mathrm{C}=\mathrm{N})$. The $\lambda_{\text {max }}(\mathrm{MeOH})$ at $(301-388 \mathrm{~nm})$ responsible for $\left(\mathrm{n} \rightarrow \pi^{*}\right)$ transition of ( N and O ) atoms and at (232-252 nm) due to $\left(\pi \rightarrow \pi^{*}\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) $(\mathrm{ppm})$ for compound (2c) shows appearance asinglet band at (7.38) due to proton of( $-\mathrm{CH}=$ ) and many signals at (7.608.34 ) due to aromatic protons ( 5 H s for phenyl ring, 4 H dd for aryl ring ). The treatment of compound ( $2 \mathrm{a}-\mathrm{d}$ ) with Thiosemicarbazide led to the formation of (3a- d). Compound ( $3 \mathrm{a}-\mathrm{d}$ ) have been identified by IR spectrum which it shows the appearance of the new two bands (asymmetric \& symmetric at ( $3466-3336 \mathrm{Cm}^{-1}$ ) and (3412-3317 $\mathrm{Cm}^{-1}$ ) and band at (1693-1710 $\mathrm{Cm}^{-1}$ ) for stretching vibration of $(\mathrm{C}=\mathrm{O}$ lactam $)$. The $\lambda_{\max }(\mathrm{MeOH})$ at (315-389 nm) responsible for ( $\mathrm{n} \rightarrow \pi^{*}$ ) transition of ( N and O) atoms and at (272-266 nm) due to ( $\pi \rightarrow \pi^{*}$ ). Compounds ( $4 \mathrm{a}-\mathrm{d}$ ) have been obtained by the reaction of hydroxyl amine hydrochloride and sodium acetate anhydrous with compounds ( $2 \mathrm{a}-\mathrm{d}$ ) in ethanol and glacial acetic acid. The reaction is followed by the appearance of the new band at ( $3201-3151 \mathrm{Cm}^{-1}$ ) due to ( $\mathrm{N}-\mathrm{H}$ ) group and disappearance the band of $(\mathrm{C}=\mathrm{O})$ for lactone ring. The $\lambda$ max $(\mathrm{MeOH})$ at $(472-387 \mathrm{~nm})$ responsible for $\left(\mathrm{n} \rightarrow \pi^{*}\right)$ transition of ( N and O ) atoms and at (281-262 nm) due to ( $\pi \rightarrow \pi^{*}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO) (ppm) for compound (4b) shows appearance asinglet band at (7.02) due to proton of (-CH-) isoxazole ring and many bands at (7.5-8.2) due to aromatic protons ( 5 H s for phenyl ring, 4 H dd for aryl ring) and other singlet band at(11.34) due to proton of $(\mathrm{NH})$ group .
The treatment of compounds ( $2 \mathrm{a}-\mathrm{d}$ ) with urea in a mixture of ethanol and acetone and $\mathrm{K}_{2} \mathrm{CO}_{3}$ led to the formation of ( $5 \mathrm{a}-\mathrm{d}$ ). The reaction is followed by show appearance of the new ( $\mathrm{C}=\mathrm{O}$ ) band at (1726-1707 $\mathrm{Cm}^{-1}$ ) due to urea moity in pyrimidine ring, and appearance a new band at
(3309-3209 $\mathrm{Cm}^{-1}$ ) due to ( $\mathrm{N}-\mathrm{H}$ ) which near from oxazole ring, and other band at ( $3273-3155 \mathrm{Cm}^{-1}$ ) due to ( $\mathrm{N}-\mathrm{H}$ ) which far from oxazole ring .The $\lambda_{\max }(\mathrm{MeOH})$ at (365289 nm ) responsible for ( $\mathrm{n} \rightarrow \pi^{*}$ ) transition of ( N and O)atoms and at (292-228 nm) due to ( $\pi \rightarrow \pi^{*}$ ). Compounds (10a,b) have been synthesized by the reaction of compounds ( $3 \mathrm{a}, \mathrm{b}$ ) with chloro ethyl acetate in ethanol .The reaction is followed by the appearance of the new ( $\mathrm{C}=\mathrm{O}$ ) band at $\left(1720,1714 \mathrm{Cm}^{-1}\right)$ for the new imidazole ring, and shows the disappearance the bands of $\left(\mathrm{NH}_{2}\right)$ group. The $\lambda_{\max }(\mathrm{MeOH})$ for compound(10a) at(354 nm) responsible for ( $n \rightarrow \pi^{*}$ ) transition of ( N and O ) atoms and at ( 269 nm ) due to $\left(\pi \rightarrow \pi^{*}\right)$. Mass Spectra for compound (10b) shows identical the experimental molecular weight with theoretical molecular weight which equal to (442 $\mathrm{g} / \mathrm{mole}$ ).
Scheme (II) Compounds (6a-d) have been synthesized by the reaction of hydrazine hydrate with compounds ( $2 \mathrm{a}-\mathrm{d}$ ) in acetic acid. The reaction is followed by the appearance of two new bands at (3311-3200 $\mathrm{Cm}^{-1}$ ) and (3211-3130 $\mathrm{cm}^{-1}$ )due to two ( $\mathrm{N}-\mathrm{H}$ )group pyrazole ring and disappearance the band of $(\mathrm{C}=\mathrm{O})$ for lactone ring. The $\lambda_{\text {max }}(\mathrm{MeOH})$ at $(305-258 \mathrm{~nm})$ responsible for $\left(\mathrm{n} \rightarrow \pi^{*}\right)$ transition of ( N and O ) atoms and at ( $259-224 \mathrm{~nm}$ ) due to
 shows appearance asinglet band at (4.63) due to proton of $(\mathrm{NH})$ group far from oxazole ring which overlapped with proton of (-CH-) pyrazole ring and many bands at(7.408.26) due to aromatic protons ( $5 \mathrm{H} \mathrm{S}, 4 \mathrm{H} \mathrm{d}, \mathrm{d}$ ) and asinglet band at (13.59) due to proton of (NH) group which near from oxazole ring. The treatment of compounds ( $6 a-d$ ) with Chloro ethyl acetate led to the formation of (7a-d). Compounds ( $7 \mathrm{a}-\mathrm{d}$ ) have been identified by IR spectrum which it show appearance of the new $(\mathrm{C}=\mathrm{O})$ band at (1749-1734 $\mathrm{Cm}^{-1}$ )for ester group . The $\lambda_{\text {max }}(\mathrm{MeOH})$ at (306-229 nm) responsible for ( $\mathrm{n} \rightarrow \pi^{*}$ ) transition of ( N and O) atoms and at ( $260-214 \mathrm{~nm}$ ) due to $\left(\pi \rightarrow \pi^{*}\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO)(ppm) for compound (7d) shows appearance atriplet band at (1.21-1.25) due to protons of $\left(-\mathrm{CH}_{3}\right)$, and aquartate band at (4.17-4.24) due to two protons of $\left(-\mathrm{CH}_{2}{ }^{-}\right.$ O ) and asinglet band at (4.68) due to two protons of (-$\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), and other singlet band at (4.98) due to proton of (NH) group which overlapped with proton of (-CH-) pyrazole ring, and many bands at (7.42-8.28) due to aromatic protons ( 5 H s for phenyl ring , 4 H dd for aryl ring) . Compounds ( $8 \mathrm{a}-\mathrm{d}$ ) have been synthesized by the reaction of Compounds (7a-d) with hydrazine hydrate in ethanol, the reaction proceeds by elimination of $\left(\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}\right)$ molecule. The reaction is followed by appearance of the new two bands (asymmetric \& symmetric) at (3319-3252 $\mathrm{Cm}^{-1}$ ) and ( $3311-3288 \mathrm{Cm}^{-1}$ ) due to $\left(\mathrm{NH}_{2}\right)$ group ,and anew $(\mathrm{C}=\mathrm{O})$ band at $\left(1683-1629 \mathrm{Cm}^{-1}\right)$ for amide which shows decrease of frequency of carbonyl .The $\lambda_{\max }$ $(\mathrm{MeOH})$ at $(385-228 \mathrm{~nm})$ responsible for $\left(\mathrm{n} \rightarrow \pi^{*}\right.$ )transition of ( N and O ) atoms and at (244-213 nm ) due to ( $\pi \rightarrow \pi^{*}$ ). Mass Spectra for compound (8c) shows identical the experimental molecular weight with theoretical molecular weight which equal to ( $367 \mathrm{~g} / \mathrm{mole}$ )
. Schiff bases ( $9 \mathrm{a}-\mathrm{d}$ ) have been obtained by reaction of compound ( 8 b ) with different aromatic aldehydes in
ethanol and glacial acetic acid .The reaction is followed by disappearance the band of $\left(\mathrm{NH}_{2}\right)$ group and appearance anew band at (3331-3150 $\mathrm{Cm}^{-1}$ ) due to ( NH amide) . The $\lambda_{\text {max }}(\mathrm{MeOH})$ at $(341-260 \mathrm{~nm})$ responsible for $\left(\mathrm{n} \rightarrow \pi^{*}\right)$
transition of ( N and O )atoms and at (229-227 nm) due to ( $\pi \rightarrow \pi^{*}$ ).





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$[\mathrm{Zn}(\mathrm{NO} 3)(\mathrm{H} 2 \mathrm{O}) 2(\mathrm{~L})] \cdot(\mathrm{NO} 3) \cdot(\mathrm{H} 2 \mathrm{O}) 2$, and $[\mathrm{Zn}(\mathrm{L})(\mathrm{OBC})(\mathrm{H} 2 \mathrm{O})]\left(\mathrm{OBC}=4,4^{\prime}\right.$-Oxybis(benzoate) $)$ ., Korean Chem. Soc.,Vol. 26, 892 ( 2005).

