

## Non-Steroidal Anti-inflammatory Drugs (NSAIDs): Synthesis of Ibuprofen, Naproxen and Nabumetone

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### Abstract:

This review paper covers the major synthetic approaches attempted towards the synthesis of some Non-Steroidal Anti-Inflammatory Drugs (Naproxen, Ibuprofen and Nabumetone)

**Key words:** synthesis, Naproxen, Ibuprofen and Nabumetone

الادوية المضادة للالتهابات الغير ستيرويدية (مضادات الالتهابات الغير ستيرويدية): تخليق الايبوبروفين، نابروكسين و نابوميون  
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### الخلاصة:

هذه المقالة تتضمن التركيب الرئيسي لتخليق الادوية المضادة للالتهابات الغير ستيرويدية (نابروكسين، ايبوبروفين و نابوميون).

**الكلمات المفتاحية:** تخليق، نابروكسين، ايبوبروفين و نابوميون

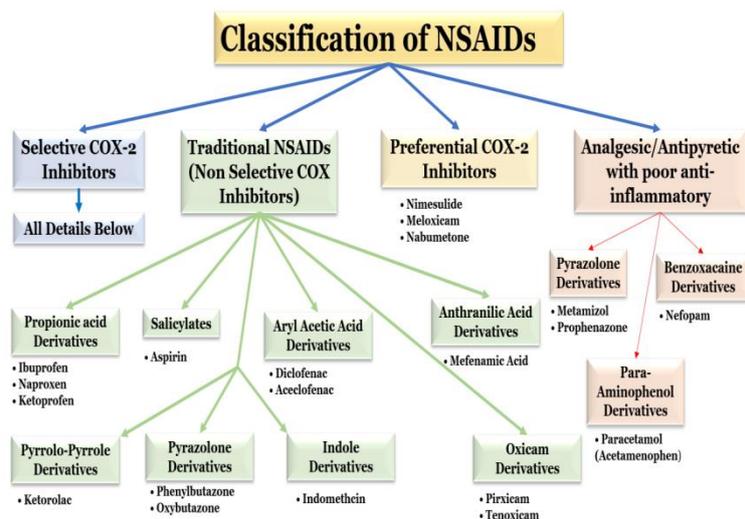
### Introduction

Non-Steroidal Anti-Inflammatory Drugs (NSAIDs) are the most prescribed therapeutic drugs for the treatment of pain and inflammation [1]. One of the major approaches to develop novel drugs is by modifying a well-known non-selective NSAIDs [2,3]. Derivatization of the carboxylate function of representative NSAIDs was shown by different studies that it could enhance the anti-inflammatory activity while minimizing the gastrointestinal side effects [4-7].

The mechanism of action principally responsible for most of the NSAIDs seems

to act by inhibition of prostaglandin (PG) synthesis causing almost complete blockade of the activity of the precursor enzymes, cyclooxygenases which are the rate limiting enzymes for Prostaglandin synthesis [8,9].

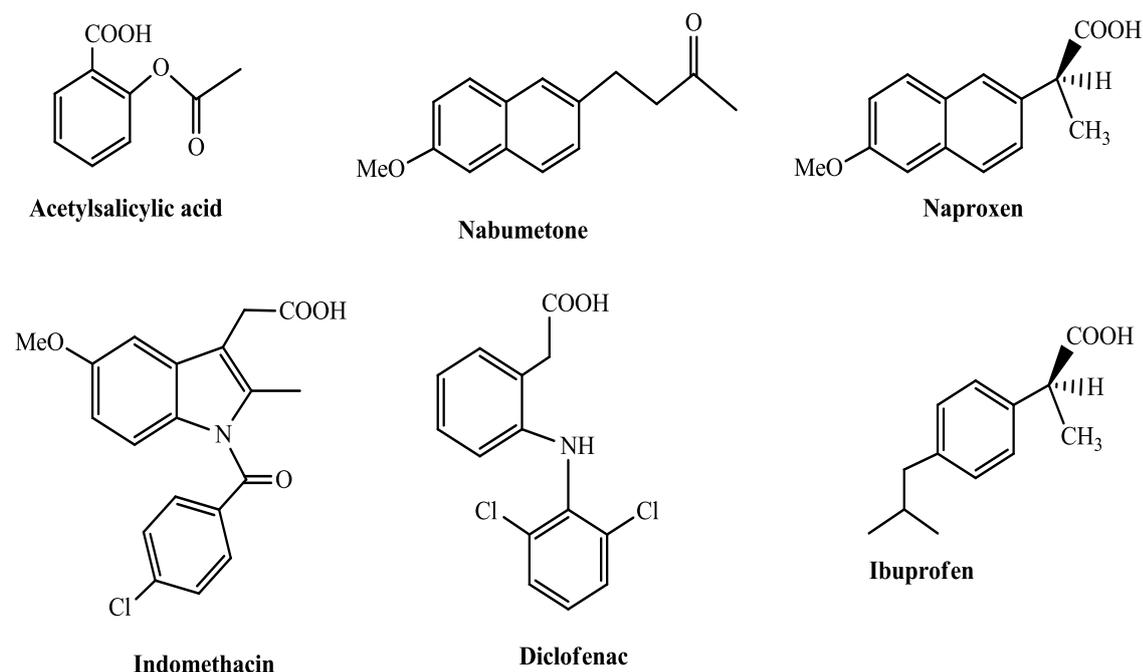
NSAIDs can be classified as traditional non-steroidal anti-inflammatory drugs, preferential COX-2 inhibitors, analgesic /antipyretic with poor anti-inflammatory NSAIDs and COX-2 selective inhibitors (Coxibs). Traditional NSAIDs can be further sub classified into a carboxylic acid, enolic acids & non-acidic compounds as shown in Figure (1) [10].



**Figure (1): Classification and representative chemical structures of the traditional nonsteroidal anti-inflammatory drugs (NSAIDs)**

Aspirin (acetylsalicylic acid) was introduced in 1899 as the first efficient drug that treats rheumatic diseases. Later on, at the time between 1960 and 1980,

various anti-inflammatory agents were developed and marketed, e.g. ibuprofen, indomethacin, diclofenac and naproxen, as shown in Figure (2) [11].

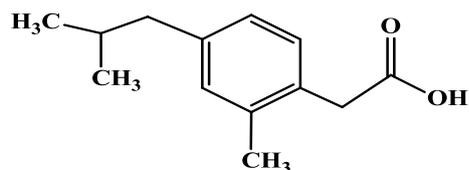


**Figure (2): Some examples of classical NSAIDs**

**Synthesis of ibuprofen:**

Ibuprofen-(+/-)2(4-iso-butyl-phenyl) also known as, propionic acid, Figure3, is considered among the most common anti-inflammatory drugs used world-wide. It's the synthetic family of 2-arylpropionic

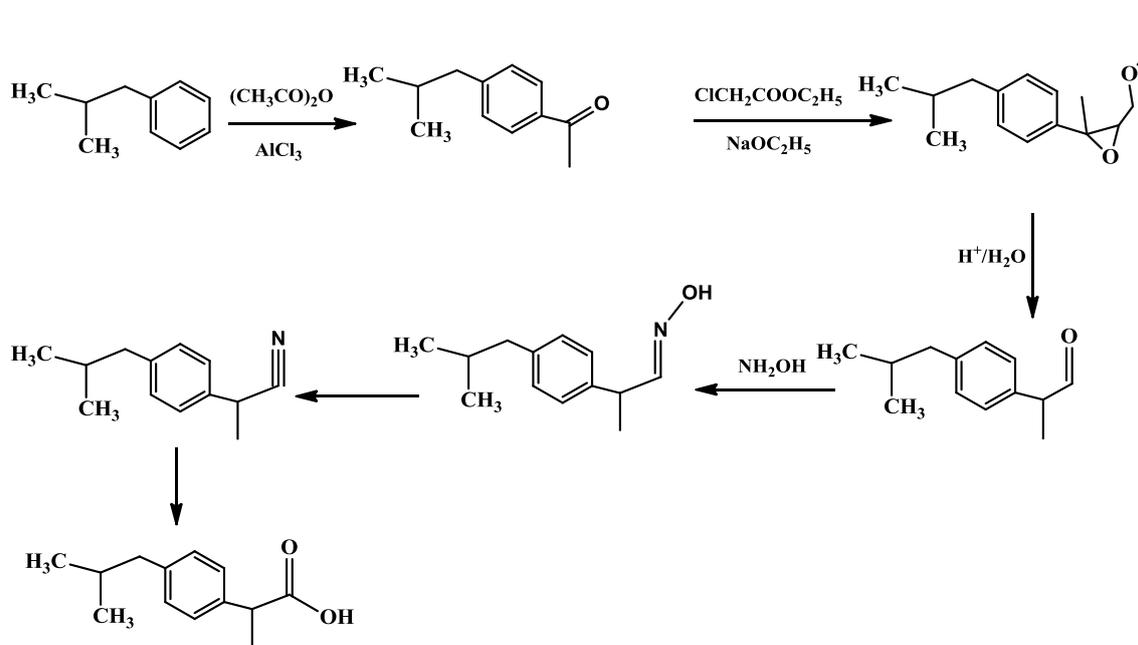
acids prototype, a subclass from Non-Steroidal Anti-Inflammatory drugs. It's not before 1960s that ibuprofen was discovered by Andrew Dunlop, whose early usage of the drug was meant as a hangover cure.



**Figure (3): Ibuprofen structure**

Boots drug company patented the procedure to synthesize the drug in the

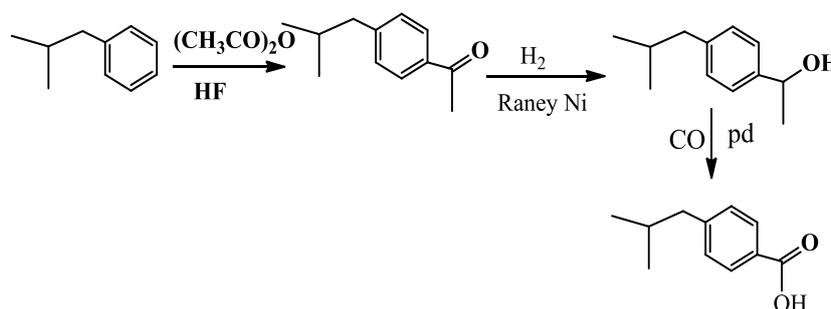
1960s Scheme (1). A chemical-waste was considerably produced as a result of applying a complicated several-step synthetic procedure [12]. Ibuprofen was discovered in 1961 by Stewart Adams and initially marketed as Brufen [13].



**Scheme (1): A synthetic method by Boots to yield Ibuprofen. The first published route to Ibuprofen.**

In 1992, Boots unearthed an alternative process of synthesizing ibuprofen Scheme

(2) that was found to be more friendly to the environment [14].



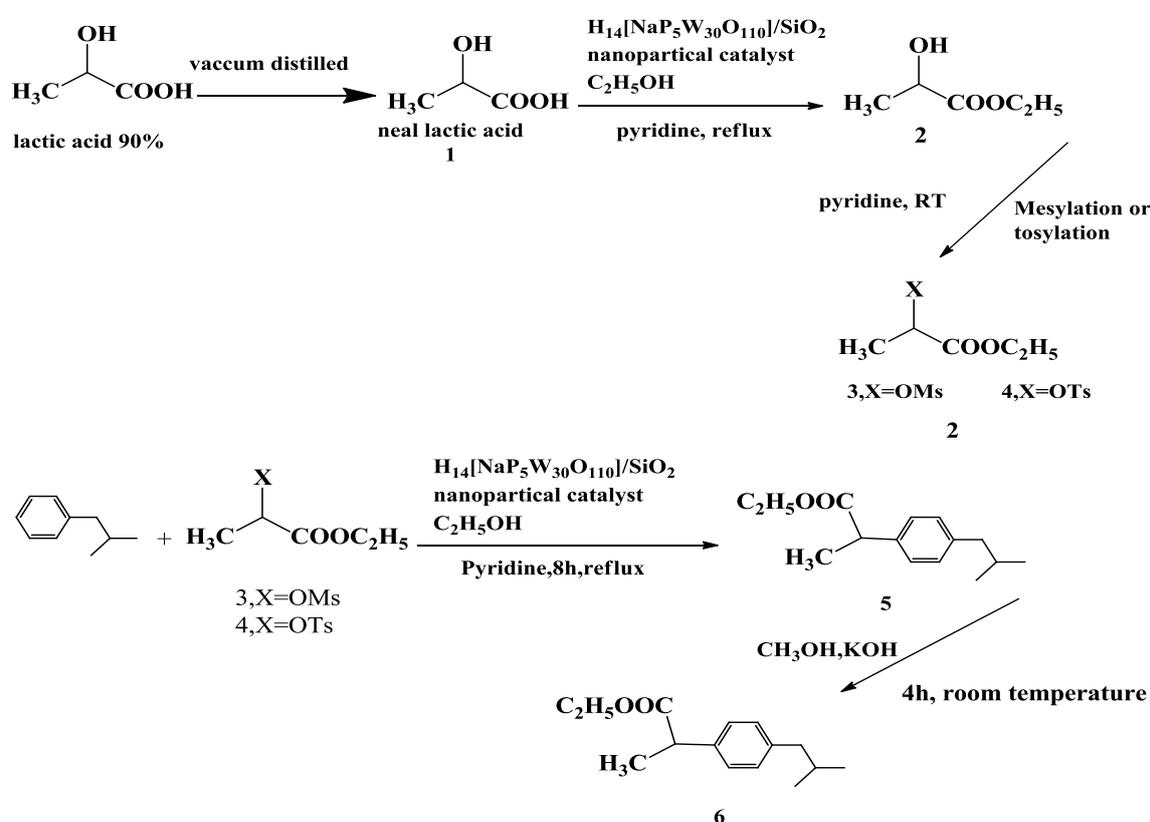
**Scheme (2): Boots procedure to synthesize ibuprofen. A green synthetic route to ibuprofen.**

Silica Preyssler Nanoparticles, also known as (H-14[Na-P5-W30-O110SiO2) (SPNPs), was used as a simple and alternative procedure to synthesize Ibuprofen. Among its many advantages it's found to be

inexpensive, efficient-catalyst and economically convenient compound. This protocol later found to produce high yields through a simple operation and easy workup procedure. Thermal-stability and

higher hydrolytic characteristic of the compound offered more advantages. The salient features of Preyssler's anion are non-toxicity, availability and reusability. Vacuum-distillation method was applied to dehydrate Lactic-acid 90% and then refluxed with ethanol in the presence of Silica Supported Preyssler Nanoparticles by azeotropic water removal to maintain 82 % ethyl-lactate (2) yield, (Scheme 1). ethyl-lactate undergo Tosylation and Methylation in the presence of pyridine or triethyl-amine at 25°C had a role in keeping the corresponding ethyl-2-

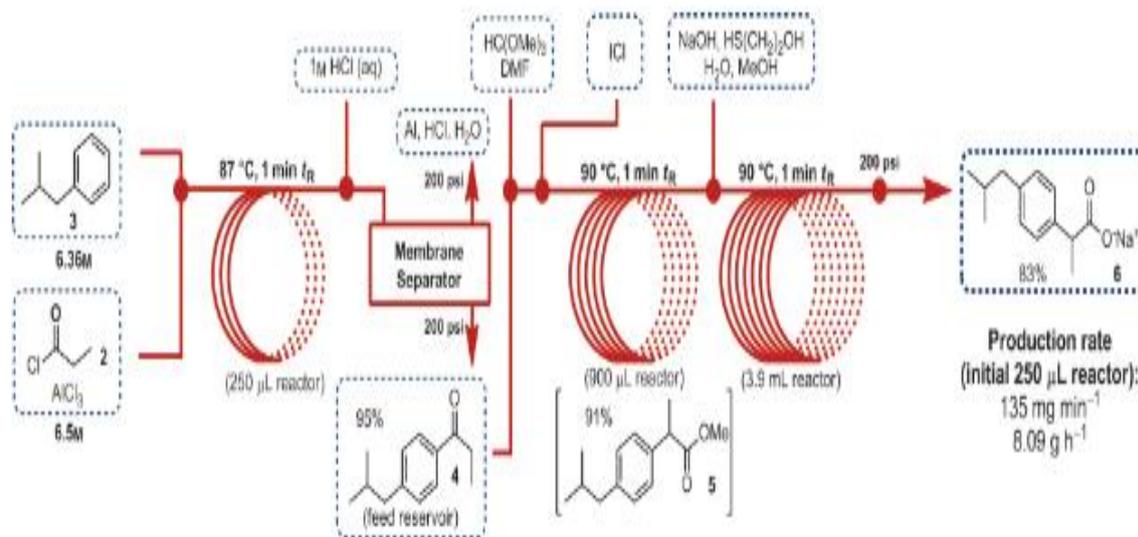
(methylsulphonyloxy) propanoate (3) and ethyl-2-(tosyloxy) propanoate (4). Friedel-Crafts-alkylation of mesylate or tosylate with isobutyl benzene for a solo step synthesis of ethyl-2-(4-isobutylphenyl) propanoate was conducted via heating with Silica-Supported -Preyssler-Nanoparticles while reaction conditions remained neat. Ethyl-2-(4-isobutylphenyl) propanoate (5) formed in 65% yield, that was hydrolyzed with KOH in methanol to obtain the racemic Ibuprofen (6) in 97% yield Scheme (3) [15].



**Scheme (3): Ibuprofen Synthesis by silica-supported-Nanoparticles (H14-[NaP5-W30-O110]/SiO2) (SPNPs)**

Flow procedures were implemented continuously to handle rough conditions in Ibuprofen production. Three operations were used sequentially; each one-step reaction was complete in a matter of a minute. A yield of 83 % was reached for the three-minute synthesis Scheme (4).

Inexpensive, simple and quickly available reagents were used like aluminum chloride and (diacetoxy) iodo-benzene with iodine-monochloride. The exothermic Friedel-Crafts reaction was conducted not only at elevated temperature, but also without external solvent [16].

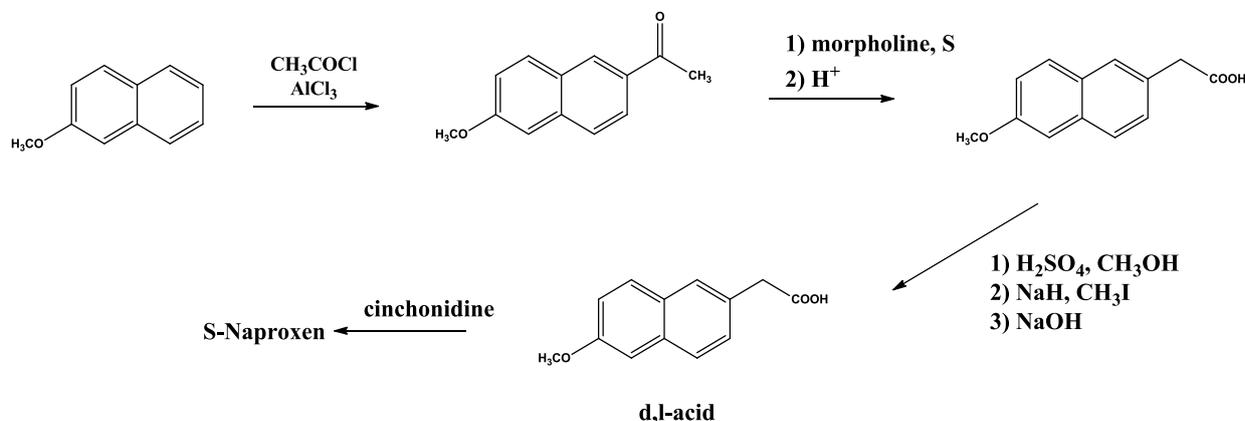


**Scheme (4): The three-minute ibuprofen synthesis**

**Synthesis of Naproxen:**

Naproxen is a member of the 2-arylpropionic acid (profen) family of NSAIDs. naproxen was produced in a large-scale-synthesis for first time in a yield of 500kg of material in 1970 Scheme (5). Friedel-Crafts-acylation of 2-

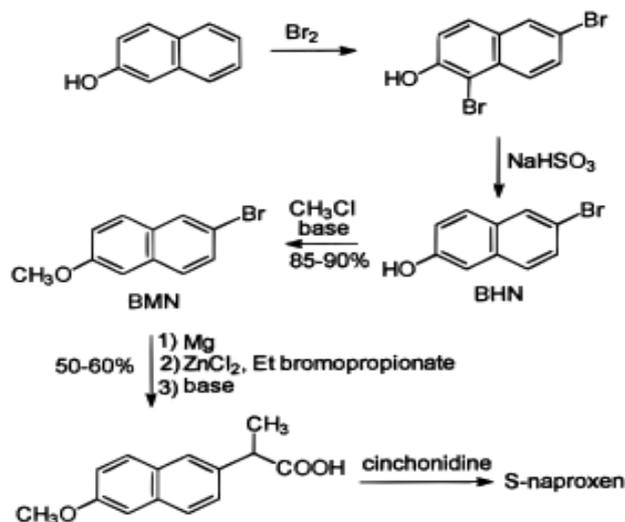
methoxynaphthalene (nerolin) supplied 2-acetyl-6-methoxynaphthalene (MAN), which in turn converted to a naphthylacetic acid via the Willgerodt reaction.  $\alpha$ -Methylation produces the d,l -acid, that can be efficiently settled using cinchonidine to synthesize S-naproxen [17].



**Scheme (5): First large-scale naproxen process (500 kg)**

The initial manufacturing step of Naproxen, around 1972-1975, was conducted through a radically different Scheme (6).  $\alpha$ -Naphthol was brominated in methylene chloride to produce 1,6-dibromonaphthol. Bisulfite was used to remove labile bromine at the 1-position. Thereafter the product of 2-bromo-6-hydroxynaphthalene (BHN) was

methylated by methyl-chloride in water-2-propanol. The resultant product of 2-bromo-6-methoxynaphthalene (BMN) found 85-90% from  $\alpha$ -naphthol. BMN was converted to a Grignard reagent, then zinc-chloride had been used to transmetalate the reagent and then naphthylzinc coupled with ethyl bromopropionate. Hydrolysis of the ester yielded d,l-acid [18].

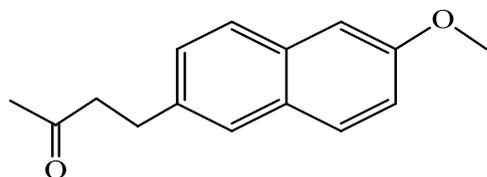


**Scheme (6): First large-scale manufacturing process, 1972-1975**

### Synthesis of nabumetone:

Nabumetone [4-(6-methoxy-2-naphthalenyl)-2-butanone] Figure (4) is a novel nonacidic broad-spectrum anti-inflammatory, analgesic, and antipyretic agent, Nabumetone was first synthesized in 1973 by Chatterjee, J. N. et al <sup>[19]</sup>.

Its activity is greater than that of aspirin and comparable to naproxen and indomethacin <sup>[20]</sup>.



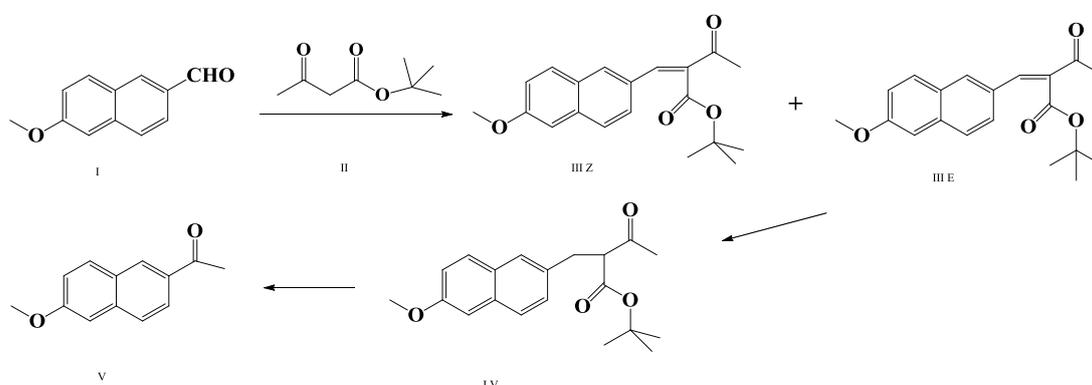
**Figure 4: Nabumetone structure**

The active metabolite of nabumetone was found to be the compound initially responsible for the therapeutic effect.

Comparatively, the parent-drug is a poor inhibitor of COX-2 by-products, particularly prostaglandins. Nonetheless, it may be less nephrotoxic than indomethacin <sup>[21]</sup>.

A process was improved to prepare Nabumetone, which consist of the next steps Scheme (7):

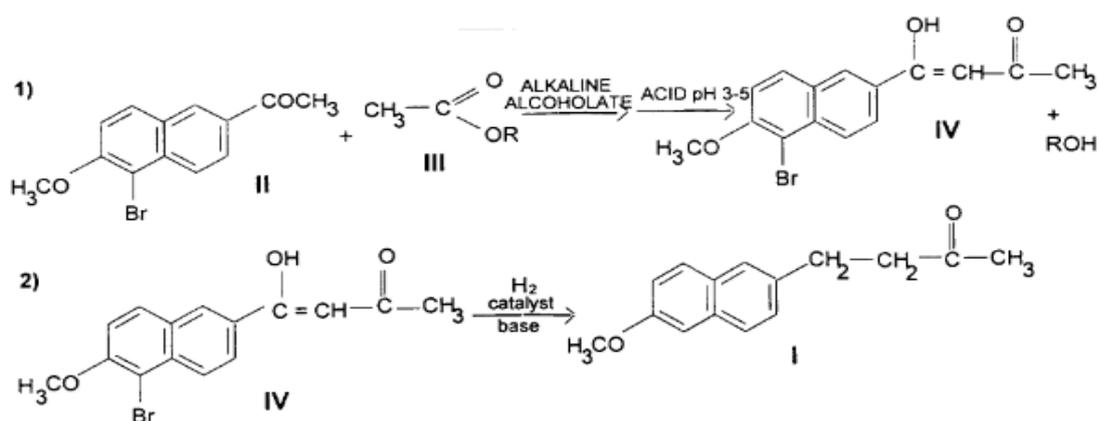
A condensation reaction between 6-methoxy-naphthaldehyde I and t-butyl acetoacetate II, to produce an E/Z mixture of 3-t-butoxycarbonyl 4-(6'-methoxy-2-naphthyl)-but-3-en-2-one III; a palladium catalyst promoted the hydrogenation of III, to gain 3-t-butoxycarbonyl-4-(6'-methoxy-2-naphthyl)-butan-2-one IV; the t-butyl ester was cleaved by acid catalysis; while the crude compound re-crystallization was achieved with methanol <sup>[22]</sup>.



**Scheme 7: synthesis of nabumetone**

Other synthetic processes of nabumetone have been described by reacting 2-acetyl-5-bromo-6-methoxynaphthalene with ethyl acetate in presence of 80% sodium hydride as an oily dispersion to gain the salt of sodium 4(5-bromo-6-methoxy-2-naphthyl)-4-hydroxybut-3-en-2-one, which is reduced to 4-(6-methoxy-2-

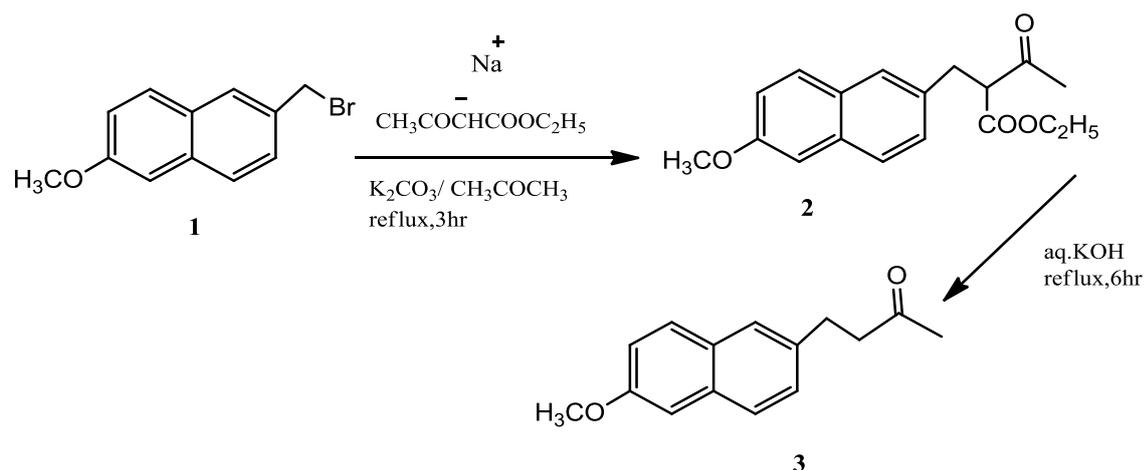
naphthyl)butan-2-one by using 10% palladium on carbon as catalyst resulting in a catalytic hydrogenation while an excess of sulfuric acid was presented with respect to the amount necessary to liberate the 4-(5-bromo-6-methoxy-2-naphthyl)-4-hydroxybut-3-en-2-one Scheme (8) [23].



Scheme (8)

Convenient method for synthesis of nabumetone from readily accessible precursor, 2-(bromoethyl)-6-methoxynaphthalene 1. Refluxing a mixture of 1 and sodium acetoacetic ester in acetone in the presence of  $K_2CO_3$  followed by usual

work up afforded ethyl-2-acetyl-3-(6-methoxy-2-naphthyl) propionate 2 as a brown syrup. Hydrolysis of 2 in presence of 40% KOH finished nabumetone in 73% yield Scheme (9) [24].



Scheme (9)

### Conclusions:

In summary, the synthesis of some Non-Steroidal Anti-Inflammatory Drugs

(Naproxen, Ibuprofen and Nabumetone) can be achieved by a number of pathways.

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